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Structure–Activity Study of Novel Tricyclic Benzazepine Arginine Vasopressin Antagonists

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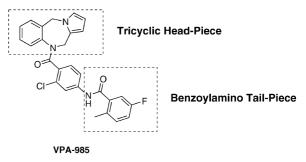
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Abstract—Novel tricyclic benzazepine derivatives were synthesized as arginine vasopressin (AVP) antagonists. Several tricyclic compounds showed potent antagonistic activity in rat AVP receptors V_{1a} and V_2 . Derivatives containing pyrrolo-tricyclic amines, **13i–k**, **30**, and **31** also showed selectivity for the V_2 receptor. © 2003 Elsevier Science Ltd. All rights reserved.

Arginine vasopressin (AVP) is a cyclic nonapeptide hormone released from the posterior pituitary in response to either increased plasma osmolality, or decreased blood volume and blood pressure. AVP binds to three known receptor subtypes: vascular V_{1a} , hormone releasing V_{1b} , and renal V_2 receptors, and regulates osmotic water permeability of water channels in the kidney through the V_2 receptor. Antagonists of the V_2 receptor are potentially useful for treating diseases characterized by excess renal reabsorption of free water, such as congestive heart failure, liver cirrhosis, nephrotic syndrome, and hyponatremia.

VPA-985 (Fig. 1), discovered in our laboratories,⁵ is an orally active AVP antagonist with selectivity for the V₂ receptor. It is currently in clinical trials for the treatment of congestive heart failure. During the course of developing VPA-985, we have investigated a large number of tricyclic benzazepine derivatives. Herein we report the synthesis and structure–activity relationship of a series of compounds containing different tricyclic 'head-pieces', as well as a series of 10,11-dihydro-5*H*-pyrrolo[2,1-*c*][1,4]benzdiazepine analogues with modifications at the benzoylamino 'tail-piece'.

We have selected eleven tricyclic amines (Fig. 2) to study the effect of various head-pieces on AVP antagonistic activity. 5,6-Dihydro-phenanthridine (1a) was prepared by reduction of the commercially available 6(5H)-phenanthridinone (Scheme 1). The 4-nitrobenzoyl derivative of 1b was synthesized according to Scheme 2. The acid chloride of 2-iodophenylacetic acid was reacted with 2-iodoaniline to give amide 2, which was reduced to amine 3 with borane-methyl sulfide. The cyclization of 3 through aryl-aryl coupling to form 1b proved to be problematic. However, the 4-nitrobenzamide 4 cyclized smoothly to 5 under copper catalyzed coupling conditions. Compound 5 was used directly to prepare the final product 13b through reduction of the nitro group, followed by acylation of the resulting amine with o-toluoyl chloride. The synthesis of 1h is



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Figure 1.

Figure 2.

Scheme 1. (a) BH₃-Me₂S, THF, reflux.

$$\begin{array}{c|c} & a,b \\ & & \\ &$$

Scheme 2. (a) SOCl $_2$; (b) 2-iodoaniline, Et $_3$ N, CH $_2$ Cl $_2$; (c) BH $_3$ -Me $_2$ S, THF, reflux; (d) 4-nitrobenzoyl chloride, Et $_3$ N, CH $_2$ Cl $_2$; (e) Cu, DMF, reflux; (f) H $_2$, 10% Pd/C, MeOH; (g) o-toluoyl chloride, Et $_3$ N, CH $_2$ Cl $_2$.

shown in Scheme 3. A mixture of 1-bromo-2-nitrobenzene and anthranilic acid was heated neat in the presence of copper and potassium carbonate to give the diarylamine 6. Hydrogenation of 6, and subsequent cyclization of the resulting amine 7 gave the lactam 8, which upon reduction with borane-methyl sulfide yielded the dibenzdiazepine 1h. Scheme 4 illustrates the synthetic routes to amines 1i and 1j. Formylation of 1-(4-nitrophenyl)-1*H*-pyrrole, followed by hydrogenation of the product 9 gave the pyrroloquinoxaline 1i.

Scheme 3. (a) Cu, K_2CO_3 , neat, $200^{\circ}C$; (b) H_2 , 10% Pd/C, MeOH; (c) xylene, reflux; (d) BH_3 –Me₂S, THF, reflux.

Scheme 4. (a) POCl₃, DMF; (b) H_2 , 10% Pd/C, MeOH; (c) ethyl oxalyl chloride, BF₃–Et₂O, CH₂Cl₂; (d) ZnI₂, NaCNBH₃, 1,2-dichloroethane, reflux; (e) SnCl₂, EtOH, reflux; (f) BH₃–Me₂S, THF, reflux.

Similarly, 1-(4-nitrophenyl)-1*H*-pyrrole was first acylated with ethyl oxalyl chloride to give compound 10, which was selectively reduced to the ester 11 with zinc iodide and sodium cyanoborohydride. Reduction of the nitro group and cyclization to the lactam was accomplished by treatment with stannous chloride in refluxing ethanol. Subsequent reduction with borane-methyl sulfide led to the pyrrolobenzodiazepine 1j.

5,6-Dihydro-11*H*-dibenz[b,e]azepine (1c),6 10,11-dihydrodibenz[b,f][1,4]oxazepine (1f),⁷ 10,11-dihydrodibenz[b,f][1,4]thiazepine (1g),⁷ and 10,11-dihydro-5*H*-pyrrolo[2,1-c][1,4]-benzodiazepine (1k)⁸ were prepared according to procedures reported in the literature. Commercially available 5,6,11,12-tetrahydrodibenz[b,f]azocine (1d) and 10,11-dihydro-5*H*-dibenz[b,f]azepine (1e) were used directly.

In order to study the structure–activity relationship of various tricyclic derivatives, the o-toluoyl group was chosen as the common tail-piece. These compounds were prepared in a convergent manner as shown in Scheme 5. Thus, the common acid chloride intermediate 12, prepared in three steps from methyl 4-aminobenzoate, was treated with the aforementioned tricyclic amines (except 1b) to give products 13a, c–k. The antagonistic activities of these compounds for the rat V_{1a} and V_2 receptors^{5c} are summarized in Table 1.

To investigate the effect of modifications of the tailpiece, we synthesized a series of analogues containing

MeO
$$NH_2$$
 a, b, c CI NH 12 d d $R-H = 1a, c-k$

Scheme 5. (a) *o*-Toluoyl chloride, Et₃N, CH₂Cl₂; (b) 1 N NaOH, MeOH; (c) SOCl₂; (d) R-H, Et₃N, CH₂Cl₂.

Table 1. Activity of tricyclic derivatives in V_{1a} and V_2 receptors

Compd	R-H	$V_{1a}\;IC_{50},\mu M^a$	V_2 IC ₅₀ , μM^a
13a	1a	0.056	0.029
13b	1b	0.098	0.025
13c	1c	0.15	0.068
13d	1d	60% (50 μM)	80% (50 μM)
13e	1e	2.5	0.86
13f	1f	1.5	1.7
13g	1g	0.40	0.86
13h	1h	0.019	0.042
13i	1i	0.17	0.066
13j	1j	0.27	0.033
13k	1k	0.038	0.004

^aBinding assays were determined by measuring the inhibition of 3 H-AVP binding to rat hepatic V_{1a} receptors or rat kidney medullary V_2 receptors.

the head-piece 1k and different substituted benzoyl groups according to Scheme 6. Acylation of 1k with 4-nitrobenzoyl chloride gave the amide 14, which was reduced by either catalytic hydrogenation or stannous chloride to the amine 15. The common intermediate 15 was then coupled with various aroyl chlorides to give the final products. Examples of these analogues and their V_{1a} and V_{2} activities are listed in Table 2.

In vivo studies were conducted on a number of the derivatives in Tables 1 and 2 using water loaded (30 mL/kg) Sprague–Dawley rats treated (ip) with AVP (0.4 μ g/kg). The aquaretic effect of these compounds was measured by the amount of urine collected after 4 h compared with control, and results are shown in Table 3.

Scheme 6. (a) 4-Nitrobenzoyl chloride, Et₃N, CH₂Cl₂; (b) H₂, 10% Pd/C, MeOH; or SnCl₂, EtOH, reflux; (d) ArCOCl, Et₃N, CH₂Cl₂.

Table 2. Activity of derivatives containing 1k in V_{1a} and V_2 receptors

Compd	X	$V_{1a}\;IC_{50},\mu M^a$	V_2 IC ₅₀ , μM^a
13k	2-Methyl	0.038	0.004
16	2-Trifluoromethyl	0.026	0.022
17	2-Methoxy	0.031	0.014
18	2-Methylthio	0.009	0.013
19	2-Ethoxy	0.045	0.077
20	2-Hydroxy	0.085	0.54
21	2-Fluoro	0.056	0.035
22	2-Chloro	0.01	0.005
23	2-Bromo	0.002	0.007
24	2,3-Dimethyl	0.24	0.013
25	2,3-Dichloro	0.027	0.029
26	2,4-Dichloro	0.023	0.003
27	2,5-Dichloro	0.12	0.014
28	2,6-Dichloro	0.094	0.015
29	2-Chloro-4-fluoro	0.007	0.004
30	2-Methyl-3-fluoro	0.026	0.004
31	2-Methyl-5-fluoro	0.02	0.0015
VPA-985	•	0.41	0.0023

^aBinding assays were determined by measuring the inhibition of ³H-AVP binding to rat hepatic V_{1a} receptors or rat kidney medullary V_2 receptors.

Majority of these compounds exhibited aquaresis activity in agreement their V_2 antagonistic activity.

It is apparent from the data in Table 1 that the tricyclic head-piece of the examples prepared has a substantial effect on their antagonistic activity for the V_{1a} and V_2 receptors. A few trends are also notable. Within the dibenz-tricyclic amine series of compounds (13a–13e) the 6,7-dihydro-5H-dibenz[b,d]azepine derivative 13b exhibited the best V_2 activity as well as selectivity over V_{1a} . Azepine head-pieces containing another heteroatom (examples 13f–13h) tend to diminish both activity and selectivity. The pyrrolo-tricyclic compounds (13I–13k) showed higher selectivity for the V_2 receptor. Compound 13k, with the 10,11-dihydro-5H-pyrrolo[2,1-c][1,4]benzdiazepine head-piece 1k, also showed potent V_2 activity.

Substitutions on the phenyl ring of the benzoyl tailpiece appear to influence selectivity for the V_2 receptor,

Table 3. Aquaretic effect in normal Sprague-Dawley rats

Compd	Dose (ip) (mg/kg)	No. of rats	Urine volume (mL/4 h)
Control (10%DMSO)		6	12.1±1
AVP control		6	2 ± 0.2
13b	10	2	13.5
13c	10	6	15.2 ± 2.4
13g	10	2	7.7
13h	30	2	7.4
13i	10	2	5.2
13j	10	2	7
13k	10	7	15.8 ± 1
16	10	2	12.5
17	10	2	9.3
22	10	2	12.4
24	10	2	18.0
26	10	8	20.4 ± 1
27	10	2	16.9
28	10	2	11.3
30	10	2	14.1
31	10	2	21.8
VPA-985	10	2	22

as data on examples containing the common head-piece 1k in Table 2 indicate. In general, a methyl group at the 2-position (13k, 30, and 31) enhances V_2 over V_{1a} selectivity significantly. Surprisingly, a 2-hydroxy substitution (20) reverses the selectivity. Among other 2-substituted analogues, the 2-chloro derivative showed better V_2 activity. However, most of them lost selectivity. The 2,4-dichloro compound 26 seems to be superior to the other dichloro-substituted positional isomers. A 3-fold increase in V_2 activity was achieved with the 5-fluoro substituted compound 31 in comparison with compound 13k, while V_{1a} activity was relatively unchanged.

In summary, we have discovered a number of novel, potent, as well as selective, AVP antagonists of the V_2 receptor by incorporating tricyclic benzazepines as head-pieces in our designed targets. Further structure–activity study of a series of derivatives containing the most promising head-piece $1\mathbf{k}$ and modifications at the tail-piece led to the discovery of compound 31, which showed potent V_2 receptor activity ($IC_{50} = 1.5$ nM) and greater than 10-fold selectivity over the V_{1a} receptor. Results from this investigation provided potential lead candidates in our effort to develop AVP antagonists for the treatment of diseases associated with excess renal reabsorption of free water.

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